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A Concise Total Synthesis of (\pm)-Ipalbidine by Application of the Aldimine-Diene Cyclocondensation Reaction

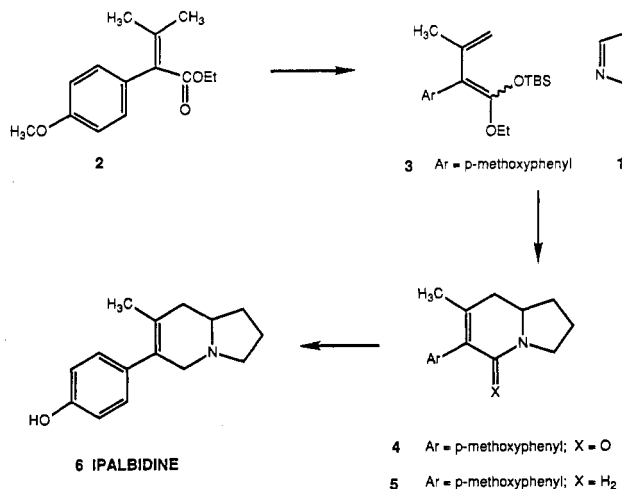
Summary: The scope of the diene-imine cyclocondensation reaction has been extended to the unstable Δ^1 -pyrroline. Cyclocondensation of this compound with the silylketene acetal derived from ethyl 2-*p*-anisyl-3-methylcrotonate provides the basis for a rapid synthesis of (\pm)-ipalbidine.

Sir: Recently we described the Lewis acid catalyzed cycloaddition of siloxydienes with aldimines.¹ Prior to our work, examples of such reactions had been restricted to specially activated imines.^{2,3} The use of highly nucleophilic dienes and Lewis acid catalysts allowed for extension of the process to a broad range of imines,¹ including dihydro- β -carboline.⁴

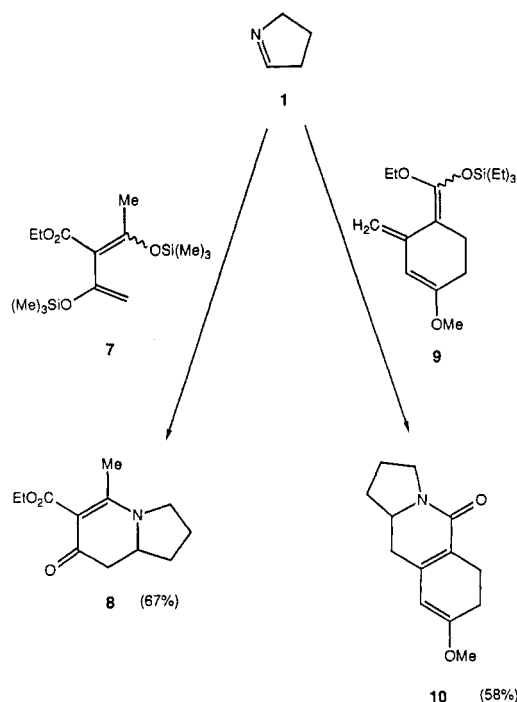
It was of interest to learn whether the parent Δ^1 -pyrroline (1)⁵ could participate in such cyclocondensations. If the reaction could be applied to imines of this type, it could have a major impact in the synthesis of various alkaloidal systems, many of which are of considerable current biological interest.⁶ We report an affirmative finding regarding this question, in the context of a particularly straightforward synthesis of ipalbidine (6).⁷

Reaction of the known α -aryl- β -methylcrotonate derivative 2⁸ with lithium diisopropylamide in THF in the presence of HMPA, followed by quenching of the resultant ester enolate with *tert*-butyldimethylsilyl chloride afforded the silylketene acetal 3 in near quantitative recovery. Reaction of 3 with 1 in methylene chloride under the influence of BF₃ etherate (-78 °C \rightarrow room temperature) affords a 40-45% yield of unsaturated lactam 4, mp 141-142 °C. Reduction of this lactam to the hexahydroindolizine 5 was accomplished (73%) through reaction with LiAlH₄-AlCl₃. Demethylation of 5 via boron tribromide in methylene chloride afforded a 78% yield of (\pm)-ipalbidine (6), mp 149-150 °C (lit.⁷ mp 149-150 °C). The assignment of the ipalbidine structure to the synthetic compound follows from comparison of its spectral properties with those previously reported.⁷

Compound 1 participates as the aldimine component with a variety of other activated dienes under Lewis acid catalysis. Two particularly interesting cases are shown.



Reaction of 1 with 7^{4,9} under the influence of zinc chloride in acetonitrile at room temperature produces vinylogous lactam 8, mp 93-95 °C in 67% yield. Similarly, reaction of 1 with silylketene acetal 9^{4,10} in chloroform at room temperature (no catalysis) gives rise to lactam 10, mp 126-128 °C in 58% yield.



These results are suggestive of a broad potential of the cyclic aldimine-diene cyclocondensation process. We are currently exploring its stereochemical ramifications in situations with substituents at the 4-position of the diene. We are also investigating the applicability of the reaction to more complex variations of compound 1. It now seems likely that this reaction will play an important role in heterocyclic synthesis.

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Supplementary Material Available: All experimental procedures and full spectral characterizations of the material described herein (3 pages). Ordering information is given on any current masthead page.

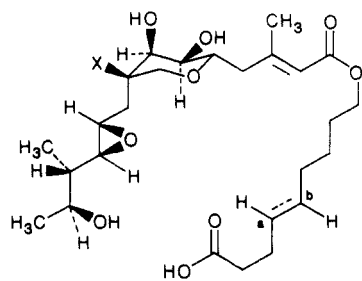
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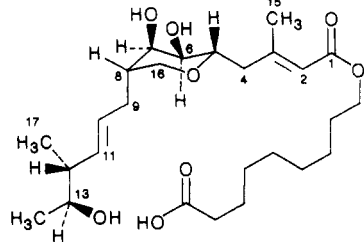
Total Synthesis of (+)-Pseudomonic Acid C[†]

Summary: The convergent enantiospecific total synthesis of (+)-pseudomonic acid C (1c) has been completed from D-glucose.

Sir: The pseudomonic acids consist of a small group of four related metabolites, originally isolated from submerged cultures of *Pseudomonas fluorescens* NCIB 10586.² Detailed studies of structural and chemical characterizations of the major component pseudomonic acid A (1a) and the lesser components B, C, and D have been reported in a series of papers.³ In 1982, X-ray crystallography established the absolute configuration of the monate skeleton of pseudomonic acid C (1c).⁴ The therapeutic value of



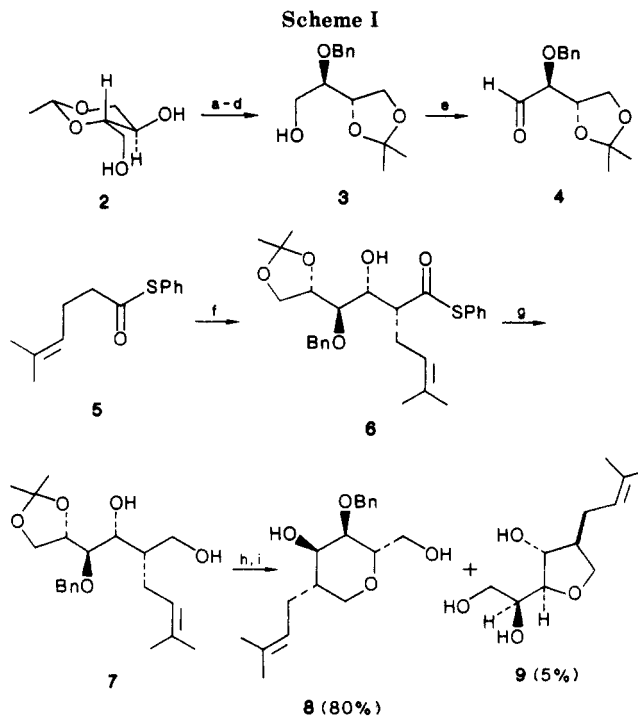
1a, X = H; C_a-C_b = C₂H₄
1b, X = OH; C_a-C_b = C₂H₄
1d, X = H; C_a-C_b = E-CH=CH



1c (pseudomonic acid C)

these antibiotics has been clinically developed in the Beecham Laboratories.⁵ Several chemical investigations have documented strategies for the total synthesis of pseudomonic acid C, and numerous approaches have been published. The conversion of pseudomonic acid C (1c) to the major metabolite 1a has also been reported.⁷ Herein, we communicate our recent efforts affording a convergent total synthesis of (+)-pseudomonic acid C.

Construction of a chiral tetrahydropyran, bearing four contiguous asymmetric centers (C₅ through C₈; 1c numbering), suggested use of a reduced (at C₁₆) carbohydrate precursor. However, the necessary absolute configuration at C₅ was directly apparent only in the series of L-hexoses.



^a (a) Trityl chloride (1.1 equiv), DMAP (1.1 equiv), DMF (80%); (b) NaH dispersion, THF, benzyl bromide (83%); (c) 40% concentrated HCl in methanol, 22 °C, 14 h (84%); (d) PPTs (0.1 equiv), acetone, 18 h (78%); (e) ClCOCOCl (1.5 equiv), Me₂SO (2.4 equiv), CH₂Cl₂, -78 °C; then Et₃N (2.4 equiv), -78 °C → 0 °C (87%); (f) 9-BBN triflate, Et₂O, thio ester 5 (1 equiv), diisopropylethylamine (1.4 equiv); add 4 at -78 °C → 0 °C (79%); (g) LiAlH₄, ether, 0 °C → 22 °C, 1 h (85%); (h) TsCl (1.4 equiv), CH₂Cl₂, DMAP (1.1 equiv), Et₃N (2.5 equiv), 22 °C, 4 h; (i) TsOH (1.1 equiv), MeOH, 16 h, 22 °C; add NaOCH₃ in methanol (80% yield of 8).

Our introduction of the desired absolute stereochemistry at C₅, as well as C₆, was accomplished by D-glucose, as illustrated in Scheme I. A classical three-step sequence provided the known diol 2,⁸ and arrangement of appropriate protection gave the optically pure primary alcohol 3. Swern oxidation provided aldehyde 4, which was used immediately for condensation at -78 °C in anhydrous ether with the preformed Z(O)-boron enolate derived from thio ester 5.^{9,10} High stereocontrol was observed with formation

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[†] Dedicated to Professor George Büchi on the occasion of his 65th birthday.